

IDENTIFYING SUBSTITUTED CYANOBUTADIYNES BY GAS-PHASE IR SPECTROSCOPY: THEORY AND EXPERIMENT

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- Cyanobutadiynes in the interstellar medium
- The technical stuff!
- IR results: experiment & theory

## **CYANOBUTADIYNES IN THE INTERSTELLAR MEDIUM**

 $R-(C=C)_2-CN$ 

If R = H:



- Cyanobutadiyne (HC<sub>5</sub>N) was detected in interstellar medium in 1976.\*
- Higher homologues have been detected up to HC<sub>11</sub>N (1997).\*\* The abundance of the cyanopolyynes decreases with length, the decrement between one to the next being about six for the longer carbon chains.



### **CYANOBUTADIYNES IN THE INTERSTELLAR MEDIUM**

- Its methyl derivative (MeC<sub>5</sub>N) was detected in that medium in 2006 in the cold dark dust cloud Taurus Molecular Cloud 1.\*
- No larger methyl derivatives have been found by now.
- Methyl derivatives can serve as indicators of gas-phase production schemes.
- Importance of Me,  $C \equiv C$ , CN groups



\*L.E. Snyder et al, *Astrophys. J.* **2006**, 647, 412 - 417.

#### **CYANOBUTADIYNES IN THE INTERSTELLAR MEDIUM**



- The bromine derivative (BrC<sub>5</sub>N) was very recently (2015) obtained by Jean-Claude Guillemin & col.\*
- $\blacktriangleright BrC_5N >> HC_7N >> MeC_7N$
- ▶ The IR spectrum of HC<sub>5</sub>N: already studied in detail.\*\*
- Our goal: to study MeC<sub>5</sub>N and BrC<sub>5</sub>N to determine the effects of substituents on the C<sub>5</sub>N group. The IR spectrum of MeC<sub>5</sub>N gives a tool for its detection and quantification.

\*N. Kerisit et al. Chem. Eur. J. 2015, 21, 6042 - 6047; \*\*Y. Benilan et al. J. Mol. Spectrosc. 2007, 245, 109 - 114

- The synthesis of these substituted cyanobutadiynes is challenging. Compounds HC<sub>5</sub>N, MeC<sub>5</sub>N are obtained by dehydration of the corresponding amide.\*
- Synthesis of BrC<sub>5</sub>N: \*\*



\*N. Kerisit et al. Chem. Eur. J. 2013, 19, 17683 - 17686;\*\*N. Kerisit et al. Chem. Eur. J. 2015, 21, 6042 - 6047

- ▶ MeC<sub>5</sub>N and BrC<sub>5</sub>N have been prepared
- ▶ IR in the 500-4000 cm<sup>-1</sup> spectral range



- Ab initio and DFT calculations MOLPRO & Gaussian09
- CCSD(T) cc-pVTZ + harmonic frequencies
  B3LYP cc-pVTZ/cc-pV5Z + harmonic frequencies Scaled & Not scaled
  B3LYP and CCSD(T) -agreement
  Best agreement with the experiment: scaled B3LYP cc-pVTZ
- **OTAIM**

M.M. Montero-Campillo et al. ChemPhysChem 2016, DOI 10.1002/cphc.201501153

- CCSD(T)
- CCSD(T) corrected \*
- MW (exp)
- QTAIM

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- Average deviation is 0.004 A
- Subtle differences on bonding that will be reflected in the IR spectra
- Rotational constant MeC<sub>5</sub>N
  B 785.13 [778.04] MHz



\* P. Botschwina. *PhysChemChemPhys* **2003**, 5, 3337-3348



 Main peaks: coupling between triple carbon bond stretching displacements and the cyano group stretching, modes (a) and (b).





- Gap between bands: BrC<sub>5</sub>N > 44.4 cm-1 (harm), 46.4 cm-1 (anharm); MeC<sub>5</sub>N > 13.7 cm-1 (harm), 23.4 cm-1 (anharm) Agreement with QTAIM
- Stretching modes Me group 3000 cm-1
- Single CC bonds stretching modes around 1228 cm-1





- BrC<sub>5</sub>N weak absorption around 760 cm-1, C-Br stretching coupled with C-C stretching modes.
- MeC<sub>5</sub>N- 1400 cm-1 deformation displacements Me group, 1000 cm-1 rocking displacements Me group.
- Some features cannot be explained considering fundamental vibrational modes: 2500 cm-1 - 1st overtone chain-stretching fundamental band (d)
- BrC<sub>5</sub>N additional band combination band involving the fundamental chain stretching band (c) and the fundamental C-Br stretching (327cm-1)



- The IR spectra of BrC<sub>5</sub>N and MeC<sub>5</sub>N have been recorded within the 4000-500 cm-1 spectral region and calculated by means of ab initio and DFT calculations.
- They look quite similar but there are subtle differences mainly in the strength of the  $C \equiv C$  bond directly attached to the substituent (distances & AIM results).
- BrC<sub>5</sub>N presents two well differentiated strong bands around 2250 cm-1, MeC<sub>5</sub>N one single band. In both cases these bands are the result of a coupling between C≡C and C≡N stretching displacements.
- The MeC5N spectrum gives a tool for its detection and quantification. The comparison with HC5N and BrC5N evidences the importance of the substituents in their spectral fingerprints.

CHEM. EUR. J. 2013, 19, 17683 — 17686; CHEM. EUR. J. 2015, 21, 6042 — 6047

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#### **SUPPLEMENTARY DATA**



\*N. Kerisit et al. Chem. Eur. J. 2013, 19, 17683 - 17686



#### **SUPPLEMENTARY DATA**



Scheme 1. Proposed mechanism of formation of  $MeC_5N$  (4) from  $HC_3N$  (1) and propyne (11).

\*N. Kerisit et al. Chem. Eur. J. 2013, 19, 17683 - 17686



Scheme 2. Proposed mechanism of formation of  $MeC_5N$  (4) from  $C_4N_2$  and: a) 1,3-pentadiyne (6), or b) propyne (11).